Total No. of Questions : 4]

P5434

SEAT No. :

[Total No. of Pages : 2

[Max. Marks : 30

[8]

[6186]-562

S.E. (Automobile & Mechanical /Mechanical S.W) (Insem) ENGINEERING THERMODYNAMICS (2019 Pattern) (Semester - III)) (Theory) (202043)

Time : 1 Hour]

Instructions to the candidates.

- 1) Answer Q.1 or Q.2, Q.3 or Q.4.
- 2) Neat duagrams must be drawn wherever necessary.
- 3) Figures to the right side indicate full marks.
- 4) Use of electronic pocket calculator is allowed.
- 5) Assume suitable data if necessary.

Q1) a) Explain in detail

- i) Macroscopic and Microscopic approach.
- ii) Path function and point function
- b) A gas having initial pressure, volume, temperature as 275kN/m², 0.09m³, 185°c is compressed at constant pressure until its temperature is 15°c. Calculate the amount of heat transferred and work done during the process. Take R = 290 kJ/kg-K C_p = 1.005 kJ/kg-K. [7]

OR

Q2) a) State "Steady Flow Energy Equation" (S.F.E.E.) and applications of steady flow energy equation on [8]

- i) Pump
- ii) Turbine
- b) A gas expands ideally, insulated nozzle following a reversible polytropic law p $v^{1.2} = C$ there is no change in potential energy but pressure drops form 20 bar to 2 bar and the specific volume increases from 0.05m³ to 0.3m³. If the entrance velocity is 80 m/s determine the exit velocity. [7]

Q3) a) State Kelvin Plank and Clausius statement of second law of thermodynamics and prove that violation of Kelvin plank statement results into violation of Clausius statement? [7]

[8]

- b) Explain
 - i) Boyles Law
 - ii) Avogadro's law
 1) Charles Law
 2) Universal gas constant
 OR

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- Q4) a) Explain Carnot cycle with PV and TS diagram give expression for the efficiency of Carnot cycle? [5]
 - b) State the limitations of first law of thermodynamics [3]
 - c) An engineer claims his engine to develop 3.75 kW. On testing the engine consumes 0.44 kg of fuel per hour having calorific value of 42000 kJ/kg. The maximum temperature recorded in the cycle is 1400 °C and minimum is 350 °C. Find whether the engineer is justified in his claim? [7]